

Name of the Student: _____

Max. Marks : 18 Marks

Time : 18 Minutes

Mark Schemes

Q1.

- | | | |
|---------|---|---|
| (a) (i) | An appropriate alkene; $\text{CH}_3\text{CH}_2\text{CHCH}_2$ or $(\text{CH}_3)_2\text{CCH}_2$ | 1 |
| | Isomer 1 | 1 |
| | Isomer 2 | 1 |
| | Position isomerism | 1 |
| | Mechanism | |
| | electrophilic attack and electron shift to Br (Unless H^+ used) | 1 |
| | carbocation | 1 |
| | reaction with carbocation | |
| | <i>[Allow mechanism marks for the alkene $\text{CH}_3\text{CHCHCH}_3$]</i> | |
| | <i>[Allow one mark if mechanism for minor product given]</i> | 1 |
| (ii) | An appropriate carbonyl; $\text{CH}_3\text{CH}_2\text{CHO}$ | 1 |
| | Mechanism nucleophilic attack and electron shift to O | 1 |
| | anion intermediate | 1 |
| | reaction with anion | |
| | <i>[Allow mechanism marks for the carbonyl $(\text{CH}_3)_2\text{CO}$]</i> | 1 |
| | Isomer 1 | 1 |
| | Isomer 2 | 1 |
| | Optical isomerism | |
| | <i>NB Isomer structures must be tetrahedral</i> | |

NB Penalise "stick" structures once in part (a)

1

- (b) QoL
Large charge on carbonyl carbon atom due to bonding
to O and Cl

1

Nucleophiles have electron pairs which can be donated

1

Equation Species

1

Balanced

1

[18]