

Name of the Student: \_\_\_\_\_

Max. Marks : 22 Marks

Time : 22 Minutes

## Mark Schemes

Q1.

- (a) **M1** The (relative) tendency of an atom to attract a pair of electrons/ the electrons/ electron density in a covalent bond

1

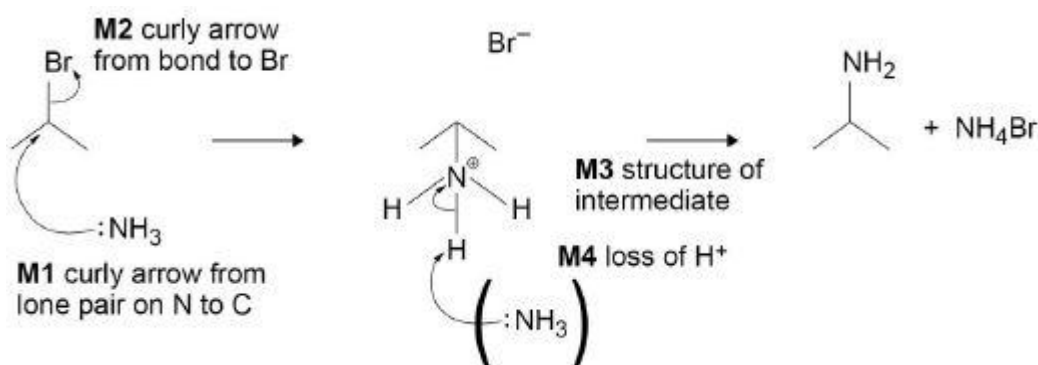
**M2** Br is more electronegative than C (or vice versa)

1

**M3** So Br is  $\delta^-$  and C is  $\delta^+$

1

(b)



**M4** Penalise loss of  $H^+$  using  $Br^-$   
 Allow  $S_n1$

4

(c) **M1**

Allow + outside square brackets

1

**M2** Use: (Hair) conditioner / (Cationic) surfactant / disinfectant  
 Allow fabric softener

1

[9]

Q2.

- (a) **M1** Q, R, S, T  
*M1 Allow the mark for candidates who correctly name or draw the isomers.*  
1
- M2** (Orange solution) turns green  
*Independent*  
1
- (b) **M1** T  
*As above*  
1
- M2** Silver mirror  
*Allow grey/black ppt*  
1
- (c) **M1** P, Q, R, S  
*As above*  
1
- M2** Sweet smelling (liquid)  
1
- M3** To react with (remove excess) acid / neutralise  
*Allow easier to identify the smell*  
1
- (d) Position  
*Allow positional*  
1
- (e) **M1** R & S have an O-H alcohols peak at 3230-3550 cm<sup>-1</sup>  
*Allow value within the range*  
1
- M2** T has C=O peak at 1680-1750 cm<sup>-1</sup>  
1
- M3** R & S (unique) fingerprint region or below 1500 cm<sup>-1</sup>  
1
- M4** Compare to a database / known spectra (and look for an exact match)  
1
- (f) All have the same  $M_r$   
*Allow*  
*same (molecular) ion M/Z peak*  
*same molecular formula*  
1

[13]