

Name of the Student: _____

Max. Marks : 21 Marks

Time : 21 Minutes

Mark Schemes

Q1.

- (a) (i)
- either**

(at terminal speed (v)) the viscous force on the droplet =
its weight (or mg or the force of gravity on it)

or

viscous force = $6\pi\eta rv$ (where r is the radius of the droplet and η is its viscosity) and
weight (= mg) = $4\pi r^3 \rho g / 3$ ✓

$$4\pi r^3 \rho g / 3 = 6\pi \eta rv \quad \checkmark$$

(which gives $r = (9 \eta v / 2 \rho g)^{1/2}$)

2

- (ii) r (can be calculated as above then) used in the
formula $m = 4 \pi r^3 \rho / 3$ to find the droplet mass, m [owtte] ✓

alternatively; (from $6\pi\eta rv = mg$) (as all values are known
use) $m = 6\pi\eta rv / g$ ✓

1

- (b) (i) electric force (or
- QV/d
-) = the droplet weight (or
- mg
-) ✓

$$Q = \frac{mgd}{v} = \frac{6.8 \times 10^{-15} \times 9.8(1) \times 5.0 \times 10^{-3}}{690} = 4.8 \times 10^{-19} \text{ C} \quad \checkmark$$

2 sf answer ✓

3

- (ii)
- any two from**

the charge on each droplet is a whole number $\times 1.6 \times 10^{-19} \text{ C}$
(or $\times$ charge of the electron) ✓

the least amount of charge (or the quantum of charge) is the
charge of the electron ✓

the quantum of charge is $1.6 \times 10^{-19} \text{ C}$ [owtte] ✓

max 2

[8]

Q2.

- (a) (i) The number of electrons (per second) in the beam will increase **(1)**
because the filament will become hotter and will emit more
electrons (per 2 second) **(1)**

2

- (ii) the speed (or kinetic energy) of the electrons will increase **(1)**

because the electrons (from the filament) are attracted towards
the anode with a greater acceleration (or force) **(1)**

(or gain more kinetic energy in crossing a greater pd)

2

- (b) (i) (magnetic) force on each electron in the beam is perpendicular
to velocity **(1)**

no work is done on each electron by (magnetic) force so ke
(or speed) is constant **(1)**

magnitude of (magnetic) force is constant because speed
is constant **(1)**

(magnetic) force is always perpendicular to velocity so
is centripetal **(1)**

max 3

- (ii) rearranging $r = \frac{mv}{Be}$ gives $\frac{e}{m} = \frac{v}{Br}$ **(1)**

$$\frac{e}{m} = \frac{7.4 \times 10^6}{6.0 \times 10^{-4} \times 68 \times 10^{-3}} = 1.81 \times 10^{11} \text{ (1) C kg}^{-1} \text{ (1)}$$

for correct answer to 2 sf **(1)**

4

- (iii) specific charge for the electron $\approx 2000 \times$ specific charge of H^+ **(1)**
(accept = and accept any value between 1800 and 2000)

which was the largest known specific charge before the specific
charge of the electron was determined/measured **(1)**

(or which could be due to a much greater charge or a much smaller
mass of the electron)

2

[13]